

0.5M NaH₂PO₄



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ACID

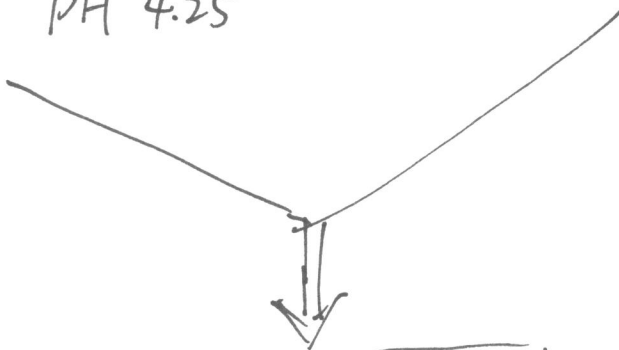
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BASE

0.5M NaH₂PO₄ · 8H₂O

pH 4.25

0.5M Na₂HPO₄

pH 9.14



A + B

pH

~~3 + 1 =~~

$$3 + 1 = 5.9$$

$$2.5 + 1 = 6$$

$$1 + 9 = 7.59$$

$$1 + 10 = 7.64$$

$$1 + 18 = 7.8$$

$$7 + 18 = 7.0$$

$$\left\{ \begin{array}{l} 7 \\ \del{1} \end{array} \right. + 6 = 7.0$$